

PAROVAYA I KARBONATNAYA KONVERSIYA METANA

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ABSTRACT

*In this work, the reactions of carbonate and steam-carbonate conversion of methane on the catalyst $(Ni_2O_3)_x * (Co_2O_3)_y * (ZrO_2)_z * (B_2O_3)_k * Me / Al_2O_3$ were studied. Depending on the temperature, the conversion of methane and carbon dioxide reaches 100%. with an increase in temperature from 700 to 800 °C, the N₂: SO ratio also increases from 1.47 to 4.00. As a result of the research, the following optimal conditions for the vapor-carbonate conversion reaction of methane in the catalyst Al₂O₃(Ni₂O₃) x * (Co₂O₃) y * (ZrO₂) z * (B₂O₃) k * Me were determined: CH₄ : CO₂ = 1:1; P = 0,1 MPa; V0 = 1000 hours⁻¹, T = 750°C.*

KEYWORDS: Methane, Carbon Dioxide, Water Vapor, Catalyst, Conversion, Synthesis Gas.

REFERENCES

1. York A.P.E., Xiao T., Creen M.L.H., Claridge J.B. Methane oxyforming for synthesis gas production // Catal. Rev. - Sci. Eng. 2007 Vol. 49. P. 511-560.
2. Bychkov V.Yu., Krylov O.V., Korchak V.N. Morozova O.S., Khomenko T.I. Investigation of the mechanism of carbon dioxide conversion of methane on Ni/α-Al₂O₃/Kinetics and catalysis. - 2002. - T. 43, no. 1. - S. 94-103.
3. Bychkov V.Yu., Tyulenin Yu.P., Krylov O.V., Korchak V.N. Carbon dioxide conversion of methane on the catalyst Co/α-Al₂O₃: formation, state and transformations of surface carbon // Kinetics and catalysis. - 2002. - T. 43, No. 5. - p. 775-782.
4. Bychkov V.Yu., Tyulenin Yu.P., Korchak V.N. Mechanism of carbon dioxide methane conversion: comparison of supported Pt- and Ni(Co)-catalysts // Kinetics and catalysis. – 2003. V. 44, no. 3. - c. 384-390.
5. Nagaoka K., Takanabe K., Aika K. Co/TiO₂ catalyst for high pressure dry reforming of methane and its modification by other metals, Stud. Surf. sci. catal. 2004 Vol. 147. P. 187-192.
6. Enger B.C., Rune L., Holmen A. A review of catalytic partial oxidation of methane to gas synthesis with emphasis on reaction mechanism over transition metal catalysts, App. catal. A., 2008. Vol. 346. P.1 - 27.
7. Fayzullaev, N. I., & Sh, S. B. (2018). Catalytic aromatization of methane with non-mo-contained catalysts. Austrian Journal of Technical and Natural Sciences, (7-8).

8. Fayzullaev, N. I., & Shukurov, B. S. (2017). Kinetics and Mechanism of the Reaction of Catalytic Dehydroaromatization of Methane. *International Journal of Oil, Gas and Coal Engineering*, 5(6), 124.
 9. Faizullaev, N. I., & Tursunova, N. S. (2018). Production of ethylene from methane using a manganese-containing catalyst. *Chemistry and Chemical Technology*, (1), 24-28.
 10. Faizullaev, N. I., & Tursunova, N. S. (2019). Kinetics of the catalytic reaction of dimerization of methane with manganese and molybdenum containing catalyst. Chief editor, 100.
 11. Fayzullayev, N. I. (2019). Kinetics and mechanism of the reaction of the catalytic oxycondensation reactionof methane. *Austrian Journal of Technical and Natural Sciences*, (5-6).
 12. Rakhmatov, S. B., & Fayzullaev, N. I. (2019). Technology for the production of ethylene by catalytycoxycondensation of methane. *European Journal of Technical and Natural Sciences*, (5-6), 44-49.
 13. Sh. B. Rakhmatov., N. I. Fayzullayev. Coke Formation of Catalyst on the Ethylene Preparation from the Oxycondensation of Methane and its Regeneration// *International Journal of Advanced Science and Technology*. Vol. 29, No. 03, (2020), pp. 7875 – 7884
 14. N. S. Tursunova., N. I. Fayzullaev. Kinetics of the Reaction of Oxidative Dimerization of Methane//*International Journal of Control and Automation* Vol. 13, No. 2, (2020), pp. 440 – 446
 15. N.I.Fayzullaev., N.S.Tursunova. Termodynamic Basis of Methane Oxidation Dimerization Reaction and Process Approval// *International Journal of Advanced Science and Technology* Vol. 29, No. 5, (2020), pp. 6522 – 6531
 16. N.I.Fayzullaev., Sh.B.Raxmatov. Kinetics and Mechanisms of OxycondensationReaction in Methane Molybden-Marganets-Zirconium Catalysis//*International Journal of Psychosocial Rehabilitation*, Vol. 24, Issue 04, 2020. ISSN: 1475-7192
 17. N.I.Fayzullayev., I.H.Ruziyev. Kinetic laws of methane carbonate conversion reaction//NEWS NEWS. 2018. 3№1. Pages 518-524.
 18. N.I.Fayzullayev., I.H.Ruziyev. Carbonate conversion of methane//NEWS NEWS. 2018. 3№1. Pages 525-532.
 19. N. I. Fayzullaev., T. B. Shoymardanov., D. R. Hamidov., 3B. Sh. Omanov. Synthesis Gas by Conversion of Methane with Carbonate Anhydride//*International Journal of Advanced Science and Technology* Vol. 29, No. 5, (2020), pp. 5991 – 6000
 20. N.I.Fayzullayev., T.B.Shoymardanov., D.M. Begimqulova., D.R.Hamidov., Kh.B. Rakhmatov. Kinetic Laws of Methane Carbonate Conversion Reaction// *International Journal of Control and Automation* Vol. 13, No. 4, (2020), pp. 268 - 276
 21. Fayzullaev N. I., Shukurov B.Sh., Normuminov A.U. Physicochemical characteristics and regeneration of the catalyst for the catalytic aromatization reaction of methane // SamSU, Scientific Bulletin. 2017, issue 1 (101). 91-98.
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- 22.** Fayzullaev N. I., Shukurov B. Sh., Normuminov A.U. Kinetics and Mechanism of the Reaction of Catalytic Dehydroaromatization of Methane // International Journal of Oil, Gas and Coal Engineering, -2017; 5(6): P. 124-129.
- 23.** Fayzullaev N.I., Shukurov B. Sh. Catalytic aromatization of methane with non-Mo-contained catalysts // Austrian Journal of Technical and Natural Sciences. July – August, № 7–8, 2018. P. 73-81.
- 24.** Fayzullaev N.I., Shukurov B. Sh., Rakhmatov Sh.B. Extraction of high-silicon zeolites from kaolin // SamSU, Scientific Bulletin. 2018, No. 5 (109), 106-111.
- 25.** Fayzullaev N.I., Shukurov B. Sh. Synthesis of high silicone zeolites and application of methane in catalytic aromating reaction // Journal of critical reviews. V. 7, issue 14, 2020. P. 1235-1242.
- 26.** N.I. Fayzullaev, B. Sh. Shukurov, A. T. Saginaev, A. Ikromov, H. Dustov. Catalytic Aromating of Methane in Mesoporous Catalysts//ISSN: 0193-4120 Page No. 26826 – 26835
- 27.** Shukurov Bakhriddin Shodikulovich. Study of the reaction of catalytic aromatization of methane.//ACADEMICIA: An International Multidisciplinary Research Journal. Impact Factor: SJIF 2020 = 7.13
- 28.** Fayzullaev N.I., Shukurov B. Sh., Saginaev A.T., Kholliev Sh.X. Kataliticheskaya degidroaromatizatsiya neftyanogo poputnogo gaza // Vestnik Atyrauskogo universiteta nefti i gaza. 2020, № 1 (53), p. 18-25.