

THE COMPOSITION OF SOAPS AND ITS EFFECT ON SOAP QUALITY

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ABSTRACT

This article provides information on the origin of soaps, the impact of changes in soap composition on changes in its quality. The women who did the laundry there noticed that the clothes were washed much better in this mixture, and gradually used this water not only for washing clothes, but also for washing the body. The carboxyl group is a polar group with a dipole moment of 1.7 D and has clearly exhibited water-like properties, including polar solvents. Attempts to introduce very fine foaming agents do not lead to the expected result because they do not stay in the soap for long and absorb it with their scent.

KEYWORDS: Soap, palmitin, stearin, olein, myristin, laurin, beef oil, etalon.

REFERENCES:

1. Azimov A. A Brief History of Chemistry. Moscow. Peace. 1983. p.145.
 2. Boyle R. The sceptical chymist. In: A.V. Tupchiev (Ed). London. 1966. p.241
 3. Jua M. History of chemistry. Moscow: Peace; 1966. p.452.
 4. Plesovskikh VA, Bezdenezhnykh AA. Physical chemistry and soap production technology. Khimizdat; 2007 . p. 336.
 5. Belov VN. Chemistry and technology of aromatic substances. Moscow: Food. Promizdat; 1983. p. 453.
 6. Bondar AG, Statyukha GA. Planning experiments in chemical technology. Kiev. Vishcha school; 1976 . p. 183.
 7. Voitkevich SA. 865 fragrances for perfumery and household chemicals. Moscow: Food. Promizdat; 1994. p. 367.
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8. Bratus IN. Chemistry of aromatic substances. 3rd ed. Moscow: Agropromizdat. 2002. p. 357.