A REVIEW PAPER ON ACIDS AND BASES

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ABSTRACT

The theory of acids and bases, like many other chemical theories, has undergone numerous changes in recent times. acids are sour in taste and change the color of blue litmus to red, whereas, bases are bitter and change the color of the red litmus to blue. Litmus is a natural indicator; turmeric is another such indicator. Acid-base indicators can be used to distinguish between an acid and a base. The universal indicator shows different colours at different concentrations of hydrogen ions in a solution. A scale for measuring hydrogen ion concentration in a solution, called pH scale has been developed. The p in pH stands for 'potenz' in German, meaning power. On the pH scale we can measure pH generally from 0 (very acidic) to 14 (very alkaline). pH should be thought of simply as a number which indicates the acidic or basic nature of a solution. Higher the hydronium ion concentration, lower is the pH value.

KEYWORDS: Acids, Bases, Salts, Hydrogen Concentration, pH Value.

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ACADEMICIA: An International Multidisciplinary Research Journal ISSN: 2249-7137 Vol. 11, Issue 12, December 2021 SJIF 2021 = 7.492 A peer reviewed journal

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